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valence electrons does a neutral magnesium  
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magnesium ion? What does magnesium have to do  
to form such an ion, and why does it tend to

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do so? 3. How many valence electrons does a single neutral chlorine (Cl atom) have? 4.

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bond is polar covalent. (b)  $\text{P} = \text{ENCl}$  "



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ENP. = 3.0" 2.1 !ENCl"P = 0.9. Since 0.5 < EN. A. < 1.7, the PCl bond is polar covalent.  
(c) !ENCl"Li Section 8.5 Electronegativity and Polarity 265 Compare and contrast polar . that is in excess.

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